# *Supplemental Material for*

# **Theoretical prediction of superconductivity in boron kagome monolayer:** *M***B<sup>3</sup> (***M* **= Be, Ca, Sr) and the hydrogenated CaB<sup>3</sup>**

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# **Content**



### **1. Computational details**

We perform the first-principles calculations within the density functional theory (DFT) to calculate the electronic structure using the Vienna ab-initio simulation package (VASP) [1] and the well-established Quantum-ESPRESSO package<sup>[2]</sup>. The Perdew-Burke-Ernzerhof generalized gradient approximation (PBE-GGA) is chosen for the exchange-correlation functional<sup>[3]</sup>. The Projector-Augmented-Wave (PAW) method is adopted to describe the electron-ion interactions. Density functional perturbation theory (DFPT)<sup>[4]</sup> is used to calculate the dynamics matrix and EPC within the range of linear response and as input to solve the isotropic Eliashberg equation. The cutoffs for wave functions and charge density are set as 80 Ry and 800 Ry, respectively. The convergence criteria for energy and force are set as  $10^{-8}$  eV and  $-0.03$  eV. The vacuum separation is set to 20 Å to avoid coupling between two neighboring monolayers. The ab-initio molecular dynamics (AIMD) simulations in the NVT ensemble are employed to investigate the thermal stability of these 2D  $MB_3$  ( $M = \text{Be}$ , Ca, Sr) monolayers. They are performed by using  $5 \times 5 \times 1$ supercells with a time step of 1 fs and total simulation of 6 ps. For the calculation of DOS, *k*-points of 45×45×1 is used. The accurate electron-phonon interaction matrix is calculated on a dense 60×60×1 *k*-point grid, and a 15×15×1 *q*-point grid is set to calculate the phonon properties and electron-phonon coupling.

The total electron-phonon coupling (EPC) constant  $\lambda$  is obtained via isotropic Eliashberg  $function^{[5-7]}:$ 

$$
\alpha^2 F(\omega) = \frac{1}{2\pi N(E_F)} \sum_{\mathbf{q}v} \delta(\omega - \omega_{\mathbf{q}v}) \frac{\gamma_{\mathbf{q}v}}{\hbar \omega_{\mathbf{q}v}} \tag{1}
$$
\n
$$
\lambda = 2 \int_0^\infty \frac{\alpha^2 F(\omega)}{\omega} d\omega = \sum_{\mathbf{q}v} \lambda_{\mathbf{q}v} \tag{2}
$$

where  $\alpha^2 F(\omega)$  is Eliashberg function and  $N(E_F)$  is the DOS at the Fermi level,  $\omega_{\mathbf{q}v}$  is the phonon frequency of the vphonon mode with wave vector  $q$ , and  $\gamma_{qv}$  is the phonon linewidth<sup>[5-7]</sup>. The  $\gamma_{\mathbf{q}\nu}$  can be estimated by

$$
\gamma_{qv} = \frac{2\pi\omega_{qv}}{\Omega_{BZ}} \sum_{k,n,m} \left| g_{kn,\,k+qm}^{\nu} \right|^2 \delta(\varepsilon_{kn} - E_F) \delta(\varepsilon_{k+qm} - E_F)
$$
 (3)

where  $\Omega_{BZ}$  is the volume of the BZ,  $\varepsilon_{kn}$  and  $\varepsilon_{k+qm}$  indicate the Kohn-Sham energy, and  $g_{kn, k+qm}^v$  represents the screened electron-phonon matrix element.  $\lambda_{qv}$  is the EPC constant for phonon mode  $q\nu$ , which is defined as

$$
\lambda_{\boldsymbol{q}\nu} = \frac{\gamma_{\boldsymbol{q}\nu}}{\pi\hbar N(E_F)\omega_{\boldsymbol{q}\nu}^2} \tag{4}
$$

 $T_c$  is estimated by McMillan-Allen-Dynes formula<sup>[7]</sup>:

$$
T_c = \frac{\omega_{\text{log}}}{1.2} \exp\left[\frac{-1.04(1+\lambda)}{\lambda - \mu^*(1+0.62\lambda)}\right] \tag{5}
$$

The hysteretic Coulomb pseudopotential  $\mu^*$  in Eq. (5) is set to 0.1 and logarithmic average of the phonon frequencies ωlog is defined as

$$
\omega_{\log} = \exp\left[\frac{2}{\lambda} \int_0^{\omega} \alpha^2 F(\omega) \frac{\ln \omega}{\omega} d\omega\right]
$$
 (6)

For the calculation of superconducting properties, the δ functions for electrons and phonons are smeared out by a Gaussian function with the broadening of 0.01 Ry and 0.16 THz, respectively.

#### **2. Lattice structure and stability of BeB<sup>3</sup> and SrB<sup>3</sup>**

The lattice structure of  $BeB_3$  and  $SrB_3$  are shown in Fig. S1 (a,b). After fully optimization, the structure parameters of BeB<sub>3</sub> and SrB<sub>3</sub> are 3.464 Å and 3.465 Å, respectively. The heights of Be and Sr to boron kagome lattice (*h*) are 0.65Å and 2.30 Å, and the B-B bond lengths are the same, i.e., 1.732 Å. The AIMD shows the variation of free energy of BeB<sub>3</sub> and SrB<sub>3</sub> within 6 ps, along with the last frame of the photographs in Fig.  $S1(c)$  and Fig.  $S1(d)$  for BeB<sub>3</sub> and SrB<sub>3</sub>, respectively. The structure [stability](file:///D:/Program%20Files%20(x86)/Youdao/Dict/8.10.3.0/resultui/html/index.html) is maintained at 400 K for BeB<sub>3</sub> with the energy fluctuates around -556.5 eV and at 220 K for SrB<sub>3</sub> with the energy fluctuates around -488.5 eV. In Fig. S1(e,f), the charge density difference clearly shows the charge transfer for BeB<sub>3</sub> and SrB<sub>3</sub>.



Fig. S1: Top and side views of the lattice structures of BeB<sub>3</sub> (a) and SrB<sub>3</sub> (b). Green (red, orange) spheres represent B (Ca, Sr) atoms and the unit cell is shown. *h* is the perpendicular distance between metal atoms and B kagome lattice. (c) and (d) The variation of the free energy in the AIMD simulations during the time scale of 6 ps along with the last frame of photographs at 400 K for BeB<sub>3</sub> and 220 K for SrB<sub>3</sub>, respectively. (e) and (f) Charge density difference for BeB<sup>3</sup> and SrB3. The color blue (yellow) represents the area where electrons are lost (obtained).

### **3. Final relaxed positions of H atoms in HCaB<sup>3</sup>**

We have considered several possible structures of hydrogenated monolayer CaB<sub>3</sub>. Within the CaB<sub>3</sub> unit cell, there are four highly symmetric positions where the H atoms can occupy. The initial positions of H atoms above the center of triangle formed by B atoms, above the B and Ca atoms, and above the B-B bonds areshown in Figs. S2(a-d). After full relaxation, the corresponding final positions of H atoms are shown in Figs. S2(e-h). For the H atoms above the B and Ca atoms, only the height of them changes after full relaxation, as seen in Fig.  $S(0)$  and  $S(0)$ . From Figs.  $S(0)$  and  $2(h)$ , it is clearly seen that H atoms move from above the B-B bonds to the hollow center of the B triangles, which is the same with that in Fig. S2(e). The calculated final total energies for the structures in Figs. S2(e-h) are -24.9, -24.6, -22.8, -24.9 eV/cell, respectively. Thus, the configuration with H atoms above the center of triangle formed by B atoms in Fig. S2(e) shows the lowest energy, and it is the most stable structure. We choose this configuration to study the superconductivity of HCaB3.



Fig. S2 (a-d) Initial positions of H atoms above the center of triangle formed by B atoms, above the B and Ca atoms, and above the B-B bonds. (e-h) Corresponding final positions of H atoms after full relaxation. Green, blue, and pink spheres represent B, Ca, and H atoms, respectively.

### **4. Electron localization function (ELF) of** *M***B<sup>3</sup> (***M* **= Be, Ca, Sr) and HCaB<sup>3</sup>**

The ELF value ranges from 0 to 1. Generally, 0 indicates very low electron density, 1 indicates strong covalent bonding or lone pair electrons, and 0.5 indicates homogeneity of electron gas. The 2D maps of ELF for  $MB_3$  ( $M = Be$ , Ca, Sr) and HCaB<sub>3</sub> are shown in Fig. S3, in which electron distributions in (001), (010) and (110) crystal plane are shown. The ELF values are about 0.7-0.9 between neighboring B atoms for these monolayers in Figs. S3 (1), (5), (9), and (13), suggesting the strong covalent bonds between B atoms. The ELF values in the metal layers in Figs. S3(2), (6), (10), and (14) are different. That is, the homogeneous gas in the Be layer in Fig. S3(2) suggests that there are strong interactions between Be-Be atoms. However, for CaB3, the ELF value in the Ca layer between the nearest Ca atoms is about 0.1-0.3 as shown in Figs.  $S3(6, 14)$ . For  $SrB<sub>3</sub>$ , the ELF value in the Sr layer between the nearest Sr atoms is about 0.1-0.2 as shown in Fig.  $S3(10)$ . Above results indicate very weak interactions between Ca-Ca and Sr-Sr atoms.

The ELF value between B atom and Be atom in Fig. S3(3) is 0.7-0.8, and the Bader charge value of Be is 0.456 *e*, showing that there is an obvious charge transfer from Be atom to the adjacent B atoms. Above two aspects together indicate the strong covalent bonds between B and Be atoms. As shown in Figs. S3(7), (11), the small ELF values between the B atoms and the Ca/Sr atoms and Bader charge analysis shows 0.728/0.624 *e* transfer from each Ca/Sr atom to the boron kagome layer, suggesting that Ca-B and Sr-B form ionic bonds.After hydrogen absorption on CaB3, as shown in Figs. S3(15, 16), the ELF values in the middle of Ca-B bonds and Ca-H bonds are about 0.1-0.2, and the Bader charge analysis shows 1.331 *e* transfer from Ca atom, including 0.628 *e* to B kagome layer and 0.703 *e* to H layer, which indicate the formation of Ca-B and Ca-H ionic bonds.



Fig. S3: 2D maps of ELF forMB<sup>3</sup> and HCaB3. (1-4) for BeB3, (5-8) for CaB3, (9-12) for SrB<sup>3</sup> and (13-16) for HCaB3. The color of red and blue refer to the highest(1.0) and lowest (0.0) value of ELF, indicating accumulation and depletion of electrons, respectively.

### **5. Electronic structure of BeB<sup>3</sup> and SrB<sup>3</sup>**

For BeB<sub>3</sub> and SrB<sub>3</sub>, some bands cross the Fermi level, supporting that BeB<sub>3</sub> and SrB<sub>3</sub> are metal shown in Figs. S4(a, b) and Figs. S4(e, f). As shown in Figs. S4(c, d), the states around the Fermi level are mainly contributed by B-p orbitals, followed by the Be-p orbitals. For SrB<sub>3</sub>, the states around the Fermi level are mainly contributed by B-*p* orbitals, followed by the Sr-*d* orbitals.



Fig. S4: Orbital-projected band structure of B (a) and Be (b) for BeB<sup>3</sup> along high-symmetry line Γ - M - K -Γ. The total DOS of BeB<sup>3</sup> and the total DOS of B and Be (c). Orbital-projected DOS of BeB<sup>3</sup> (d). (e) - (h) are the similar with  $(a) - (d)$  but for SrB<sub>3</sub>.

#### **6. Electron-phonon coupling and possible superconductivity of BeB<sup>3</sup> and SrB<sup>3</sup>**

In the unit cell, there are four atoms leading to twelve phonon bands, which include three acoustic phonon modes and nine optical phonon modes. The range of frequency extending up to about 1214 cm<sup>-1</sup> for BeB<sub>3</sub> and 1048 cm<sup>-1</sup> for SrB<sub>3</sub>, respectively. Phonon dispersion weighted by the vibration modes of B,  $M(M = Be, Sr)$  atoms of BeB<sub>3</sub> and SrB<sub>3</sub> are shown in Figs. S5(a) and S5(e). For BeB<sub>3</sub>, the region with the largest  $\lambda_{qv}$  is in the low-frequency shows in Fig. S5(b) and in Fig.  $S5(d)$ . Eliashberg spectral function of BeB<sub>3</sub> indicates that its EPC mainly comes from the coupling between electrons and the low-frequency out-of-plane vibrations of B and Be atoms. For SrB<sub>3</sub>, the region with the largest  $\lambda_{qv}$  is also in the low frequency shows in Fig. S5(f) and in Fig. S5(h). Eliashberg spectral function of SrB<sup>3</sup> indicates that its EPC also mainly comes from the coupling between electrons and the low-frequency out-of-plane vibrations of B and Sr atoms. Particularly, the in-plane vibration of B also contributes greatly, as seen from the strong EPC  $\lambda_{qv}$  at the soften phonon spectrum along Γ-M direction.



Fig. S5: Phonon dispersion weighted by the vibration modes of the atoms for BeB3 (a) and SrB3 (e). Phonon dispersion weighted by the magnitude of EPC  $\lambda_{qv}$  for BeB<sub>3</sub> (b) and SrB<sub>3</sub> (f). Atom-projected phonon DOS for BeB<sub>3</sub> (c) and SrB<sub>3</sub> (g). Eliashberg spectral function  $\alpha^2F(\omega)$  and cumulative frequency dependence of EPC  $\lambda(\omega)$  of BeB<sub>3</sub> (d) and  $SrB<sub>3</sub>$  (h).

### **7. The vibration modes of**  $MB_3$  **(** $M = Be$ **, Ca, Sr) and**  $HCaB_3$  **at**  $\Gamma$  **point**

The modes at  $\Gamma$  point for BeB<sub>3</sub>, CaB<sub>3</sub>, SrB<sub>3</sub>, and HCaB<sub>3</sub> are shown in Figs. S6-S9. Figs. S6-S9 (1-3) show the acoustic phonon modes at zero frequency. The direction of each acoustic vibration modes are same with the same amplitude. The rests of vibration modes are optical phonon modes, revealing the relative vibrations of B and metal atoms. More information about the vibration modes at Γ point, including the symmetry, vibration, Raman/infrared activity and frequencies for BeB3, CaB3, SrB3, HCaB<sup>3</sup> are listed in Table S1-S4, which provide a theoretical basis for future experimental verification.



Fig. S6: The twelve vibration modes of pristine BeB<sub>3</sub> at Γ point are in the Brillouin zone. The vibration frequencies are shown below each figure. The rad and green spheres represent Be and B atoms, respectively, and the purple arrow represents the direction of atoms vibration.



Fig. S7: The twelve vibration modes of pristine CaB<sub>3</sub> at Γ point are in the Brillouin zone. The vibration frequencies are shown below each figure. The blue and green spheres represent Ca and B atoms, respectively, and the purple arrow represents the direction of atoms vibration.



Fig. S8: The twelve vibration modes of pristine SrB<sup>3</sup> atΓ point are in the Brillouin zone. The vibration frequencies are shown below each figure. The orange and green spheres represent Sr and B atoms, respectively, and the purple arrow represents the direction of atoms vibration.



Fig. S9: The fifteen vibration modes of pristine HCaB<sup>3</sup> at Γ point are in the Brillouin zone. The vibration frequencies are shown below each figure. The blue, pink, and green spheres represent Sr, H and B atoms, respectively, and the purple arrow represents the direction of atoms vibration.

Modes	Symmetry	Vibration	Activity	Freq. $(cm^{-1})$
1	A <sub>1</sub>	Translational	$I+R$	$\theta$
$2 - 3$	$E_1$	Translational	$I+R$	$\boldsymbol{0}$
$4 - 5$	E <sub>2</sub>	Out-plane B and Be	R	155
6	A <sub>1</sub>	Out-plane B and Be	$I+R$	463
$7 - 8$	$E_1$	In-plane B and Be	$I+R$	814
9	B <sub>2</sub>	In-plane B	$\overline{\phantom{a}}$	816
$10 - 11$	$E_1$	In-plane B and Be	$I+R$	944
12	$B_1$	In-plane B	$\overline{\phantom{0}}$	1108

TABLE S1: The symmetry, vibration, activity (I is Infrared, and R is Raman) and frequency ( cm<sup>-1</sup>) for the 12 vibration modes at  $\Gamma$  point of BeB<sub>3</sub>.

TABLE S2: The symmetry, vibration, activity( I is Infrared, and R is Raman) and frequency ( cm<sup>-1</sup>) for the 12 vibration modes at  $\Gamma$  point of CaB<sub>3</sub>.

Modes	Symmetry	Vibration	Activity	Freq. $(cm^{-1})$
$\perp$	A <sub>1</sub>	Translational	$I+R$	$\theta$
$2 - 3$	$E_1$	Translational	$I+R$	$\theta$
$4 - 5$	$E_1$	In-plane B and Sr	$I+R$	165
6	A <sub>1</sub>	Out-of-plane B and Ca	$I+R$	289
$7 - 8$	E <sub>2</sub>	Out-of-plane B and Ca	R	316
9	B <sub>2</sub>	In-plane B	$\overline{\phantom{a}}$	432
$10 - 11$	$E_1$	In-plane B	$I+R$	884
12	$B_1$	In-plane B		1136

TABLE S3: The symmetry, vibration, activity (I is Infrared, and R is Raman) and frequency ( cm<sup>-1</sup>) for the 12 vibration modes at  $\Gamma$  point of SrB<sub>3</sub>.



Modes	Symmetry	Vibration	Activity	Freq. $(cm^{-1})$
1	A <sub>1</sub>	Translational	$I+R$	$\theta$
$2 - 3$	E	Translational	$I+R$	$\boldsymbol{0}$
$4 - 5$	E	In-plane B, Ca and H	$I+R$	152
6	A <sub>1</sub>	Out-of-plane B, Ca and H	$I+R$	313
$7 - 8$	E	Out-of-plane B and Ca,	$I+R$	343
		In-plane H		
9	A <sub>2</sub>	In-plane B		438
$10 - 11$	E	In-plane B and H	$H+R$	882
12	A <sub>1</sub>	Out-of-plane H	$I+R$	893
$13 - 14$	E	In-plane H	$I+R$	902
15	A <sub>1</sub>	In-plane B, Out-plane H	$I+R$	1096

TABLE S4: The symmetry, vibration, activity (I is Infrared, and R is Raman) and frequency ( cm<sup>-1</sup>) for the 15 vibration modes at Γ point of HCaB3.

#### **8. T<sup>c</sup> of BeB3, CaB3, SrB<sup>3</sup> , and HCaB<sup>3</sup> as a function of Coulomb pseudopotential μ** ∗

The Coulomb pseudopotential  $\mu^*$  is a parameter closely related to  $T_c$  [8], which are shown in Eq. (5). Here, the relationship between Tc and  $\mu^*$  of BeB<sub>3</sub>, CaB<sub>3</sub>, SrB<sub>3</sub>, and HCaB<sub>3</sub> are studied in detail by setting  $\mu^*$  in the range of 0.05-0.15. As shown in Fig. S10, for BeB<sub>3</sub>, CaB<sub>3</sub>, SrB<sub>3</sub>, and HCaB<sub>3</sub>, T<sub>c</sub> monotonically decreases with the increase of  $\mu^*$ . For BeB<sub>3</sub>, CaB<sub>3</sub> and SrB<sub>3</sub>, when  $\mu^*$ increases from 0.05 to 0.15,  $T_c$  decreases from 6.7 K to 1.1 K, 27.2 to 17.7 K, 24.5 K to 17.5 K, respectively. The trend of  $T_c$  for HCaB<sub>3</sub> is the same as that of MB<sub>3</sub>, with  $T_c$  of 45.5 K at  $\mu^* = 0.05$ and  $T_c$  of 33 K at  $\mu^* = 0.15$ . In this paper, we choose  $\mu^* = 0.1$ , and the  $T_c$  of BeB<sub>3</sub>, CaB<sub>3</sub>, SrB<sub>3</sub> and HCaB<sup>3</sup> are 3.2, 22.4, 20.9 and 39.3 K, respectively.



Fig. S10: Evaluated T<sub>c</sub> of our studied two monolayers as a function of Coulomb pseudopotential  $\mu^*$  . Vertical line marks the value  $\mu^*=0.10$  used in this work.

# 9. Convergence of EPC strength  $\lambda$  and  $T_c$  for BeB<sub>3</sub>, CaB<sub>3</sub>, SrB<sub>3</sub>, and HCaB<sub>3</sub>

We plot the calculated EPC strength  $\lambda$  and  $T_c$  as a function of the Gaussian broadening for electrons, which are shown in Figs. S11(a) and S11(b). It is seen that both  $\lambda$  and  $T_c$  converge for the four materials. We should determine the degauss at which  $\lambda$  becomes flat, and the  $\lambda$  at this degauss should be chosen as the convergence data<sup>[9-10]</sup>. The converged values of  $\lambda$  for BeB<sub>3</sub>, CaB<sub>3</sub> SrB<sub>3</sub>, and HCaB<sub>3</sub> are 0.49, 1.09, 1.33, and 1.39 at degauss of 122.4, 122.4, 204, and 68 meV, with a corresponding  $T_c$  of 3.2, 22.4, 20.9 and 39.3 K, respectively.



Fig. S11.  $\lambda$  (a) and T<sub>c</sub> (b) as a function of degauss for pristine BeB<sub>3</sub>, CaB<sub>3</sub>, SrB<sub>3</sub>, and HCaB<sub>3</sub>.

# **10. Dynamical unstability of 2D MgB<sup>3</sup>**

We also calculate the phonon spectrum of MgB<sub>3</sub>. Fig. S12 shows that it has an obvious imaginary frequency around M point, indicating that it is dynamically unstable.



Fig. S12: Phonon dispersion of MgB3.

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