

Supplementary Information for “Partially Diffusive Helium-Silica Compound under High Pressure”

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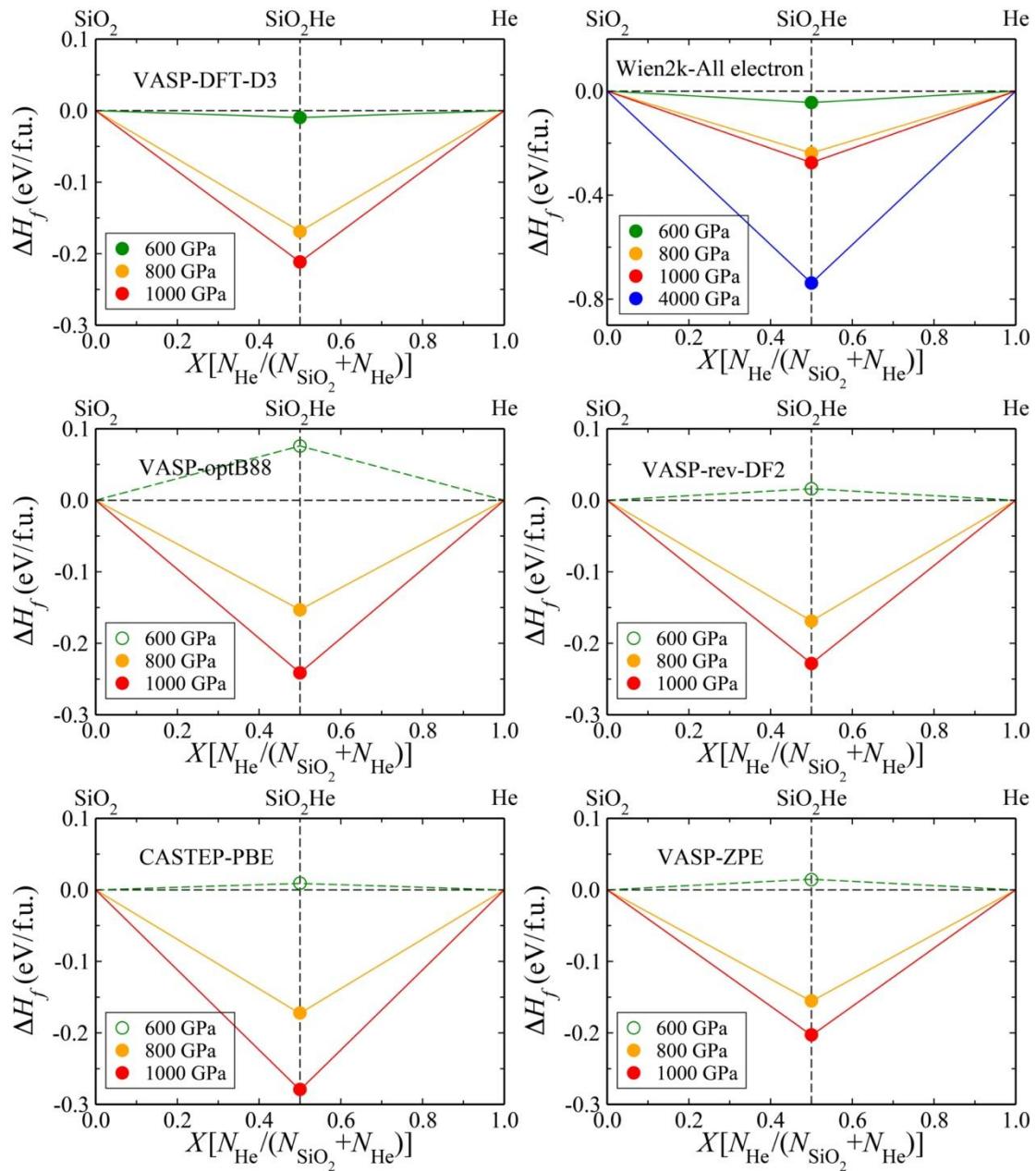


Fig. S1. Convex hull of formation enthalpies (ΔH_f) calculated from DFT-D3¹, all electron potentials (Wien2k²), vdw-optB88³ and rev-vdw-DF2⁴, CASTEP⁵, and with zero point energy (ZPE). The results are robust and consistent with different functionals and methods.

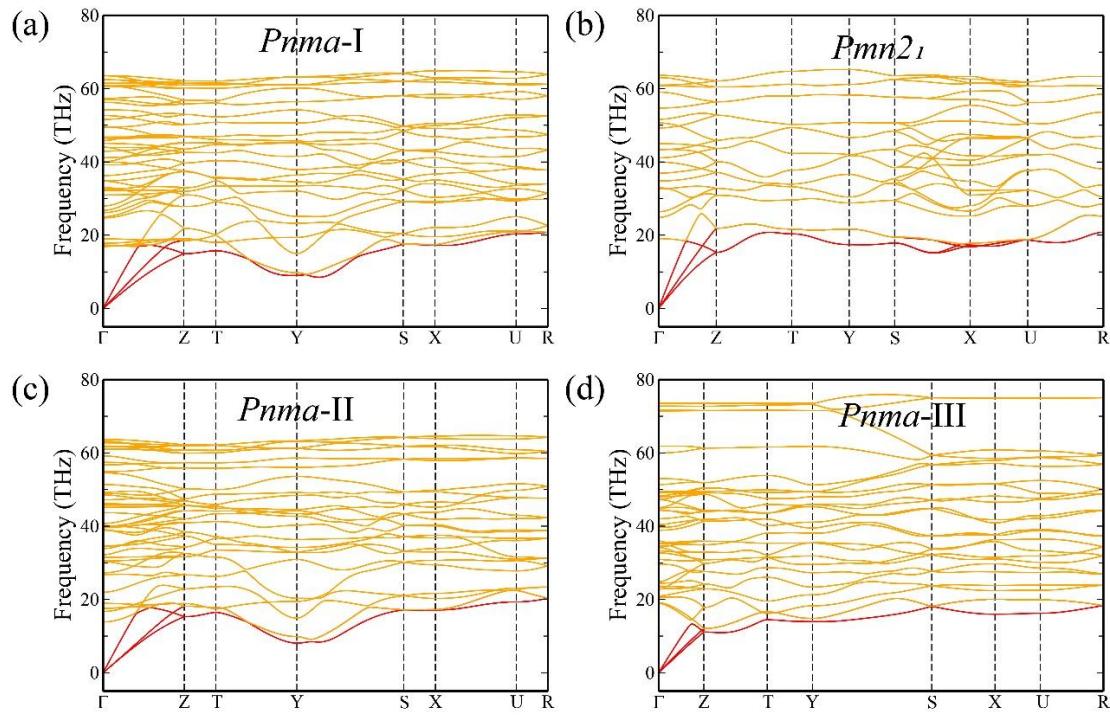


Fig. S2. Calculated Phonon spectra of the thermodynamically stable/metastable phases: (a) *Pnma*-I, (b) *Pmn2*₁, (c) *Pnma*-II, and (d) *Pnma*-III. The orange curves correspond to optical branches and the red curves correspond to acoustic branches. No imaginary vibrational frequencies were found in the phonon spectra.

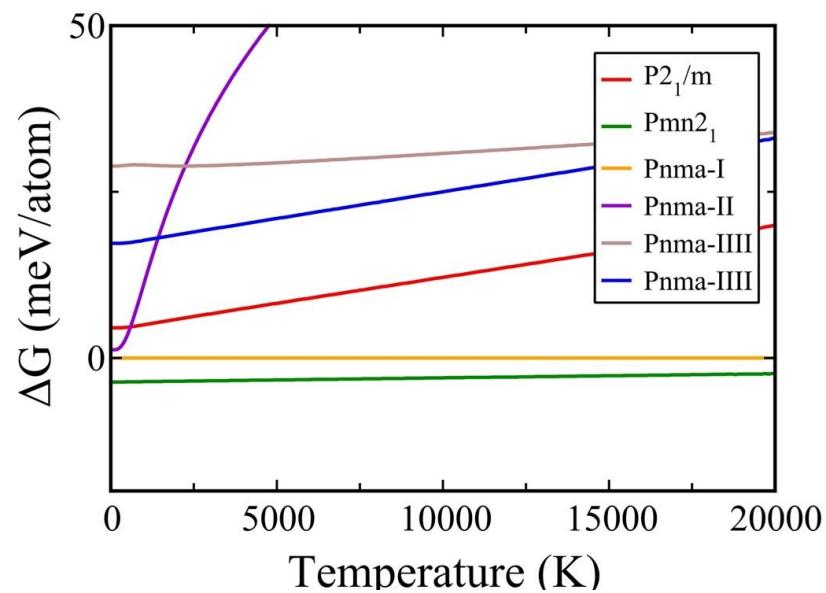


Fig. S3. Gibbs free energy at 2000 GPa based on QHA calculations.

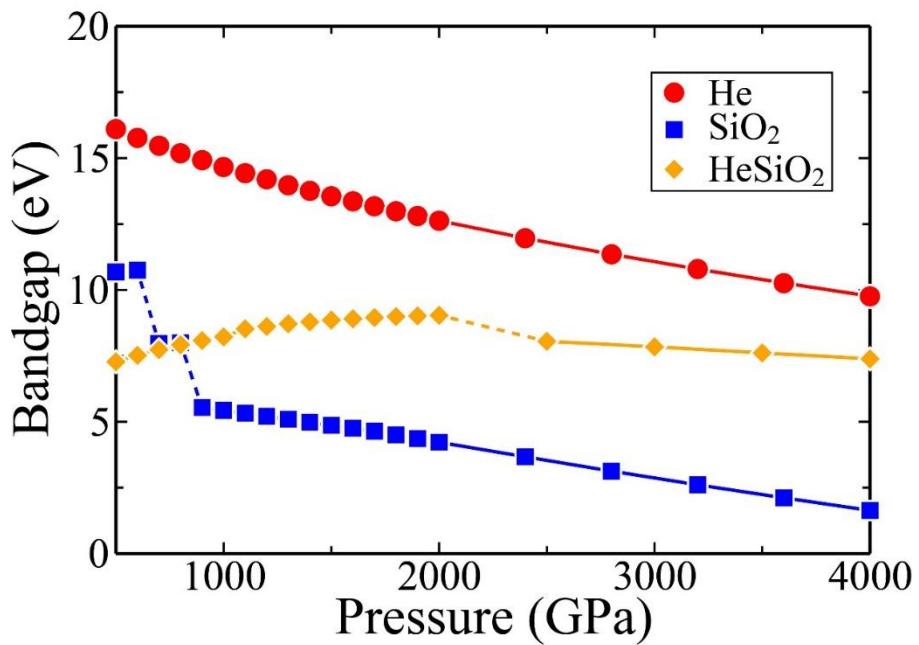


Fig. S4. Calculated Band gaps of He, SiO₂, and HeSiO₂ over the pressure range 500-4000 GPa based on simply pure PBE functional.

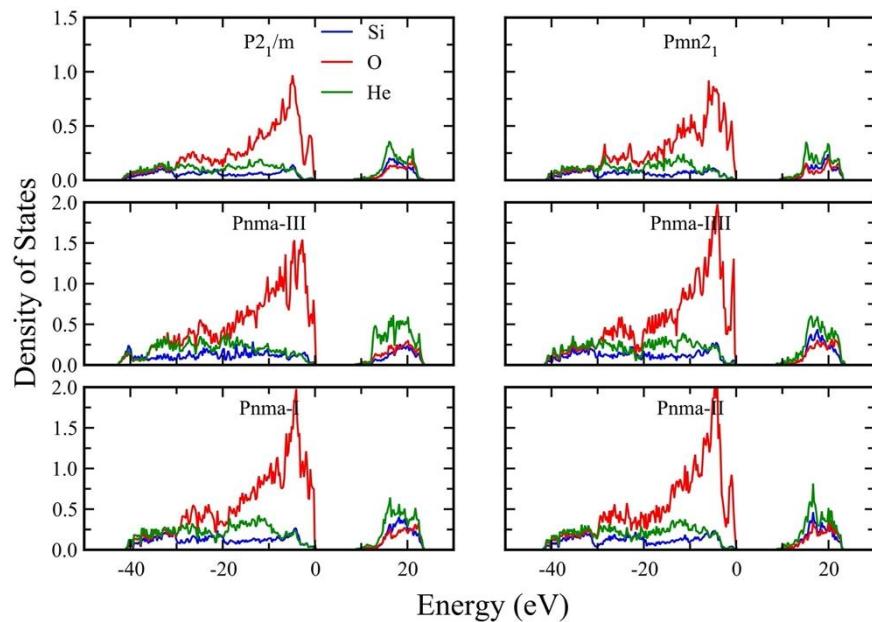


Fig. S5. Density of states for all HeSiO₂ canidates at 2000 GPa based on simply pure PBE functional.

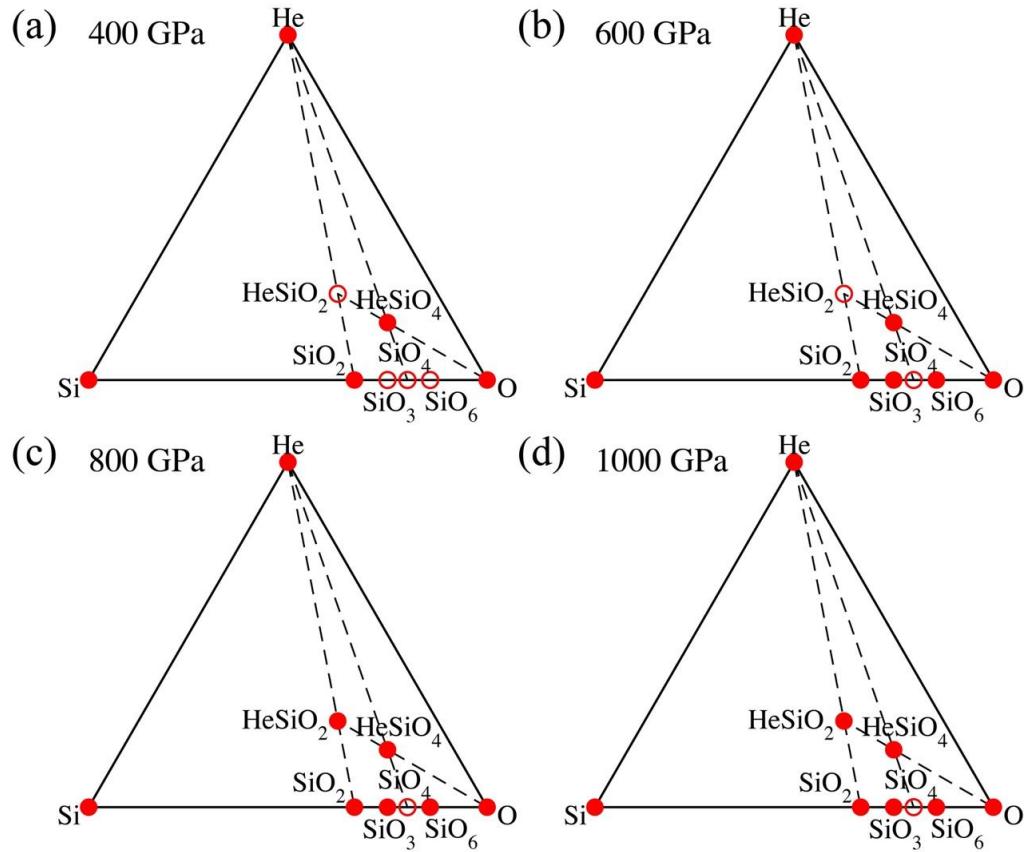


Fig. S6. Ternary phase diagram of the He-Si-O system at 400, 600, 800, and 1000 GPa. The stable/unstable phases are marked in solid/open red points.

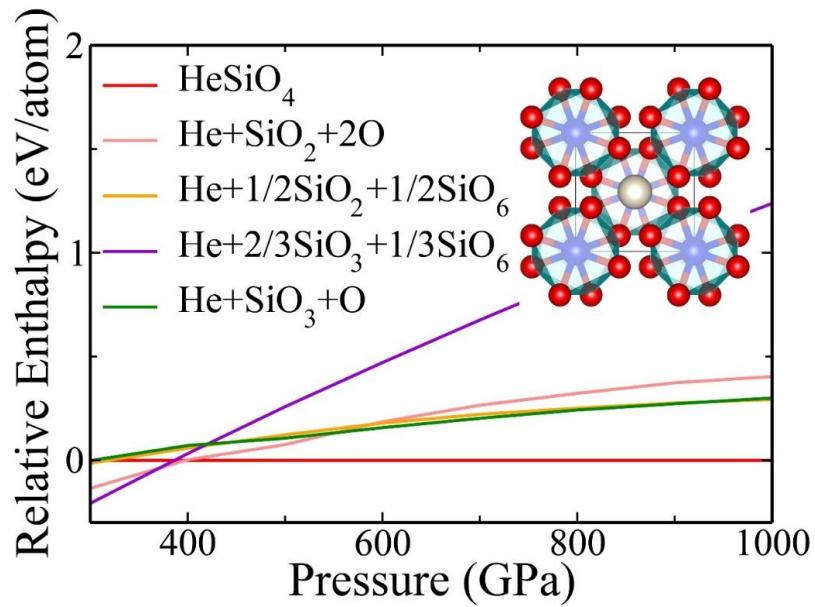


Fig. S7. Enthalpies of the HeSiO₄ compounds, as well as its possible decompositions in the pressure range between 300 and 1000 GPa. Insert figure show the crystal structure of *I*422 phase HeSiO₄.

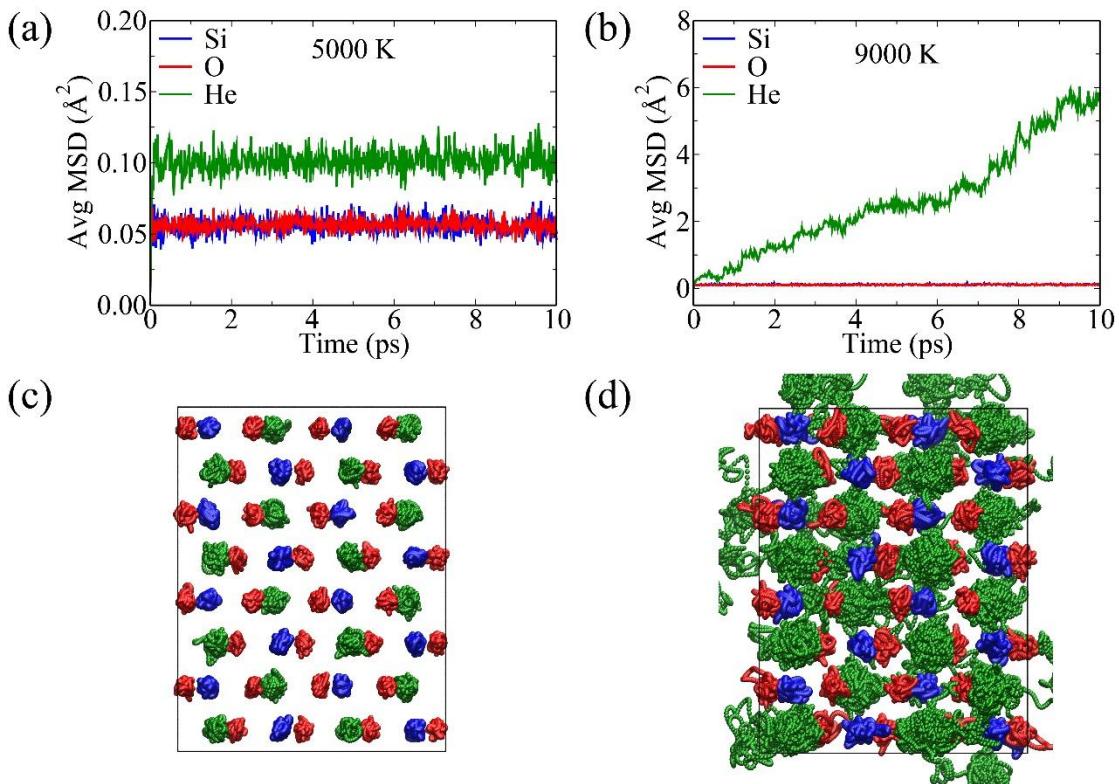


Fig S8. Dynamical behavior of Si (blue), O (red), and He atoms in the HeSiO_2 compound under high pressures and high temperatures. AIMD simulations were performed at initial pressure of 600 GPa for the *Pnma* HeSiO_2 compound. (a-b) The averaged mean-squared displacements (MSD). (c-d) Atomic trajectories from the simulations representing the plastic phase. The Si, O and He atoms are respectively plotted with blue, red, and dark green.

Table I. Bader charge on helium atoms in different phases.

Pressure	P2 ₁ /m	Pmn2 ₁	Pnma-III	Pnma-IV	Pnma-I	Pnma-II
500 GPa	2.075	2.078	2.072	2.079	2.078	2.075
1000 GPa	2.088	2.085	2.069	2.090	2.083	2.087
2000 GPa	2.095	2.083	2.056	2.089	2.088	2.093
3000 GPa	2.105	2.079	2.064	2.106	2.082	2.083
4000 GPa	2.112	2.090	2.074	2.116	2.092	2.079

Table II. Thermodynamic data from our simulations affect by different box size (number of atoms in supercells: N) and k -mesh on motion behaviors of each species of atom in *Pnma-I* phase.

V (Å ³)	T (K)	P (GPa)	N	k -mesh	D_{Si} (10 ⁻⁸ m ² /s)	D_O (10 ⁻⁸ m ² /s)	D_{He} (10 ⁻⁸ m ² /s)	State
659.04	5000	663.84	192	$1 \times 1 \times 1$	0.00	0.00	0.00	Solid
659.04	5000	663.97	192	$2 \times 2 \times 2$	0.00	0.00	0.00	Solid
1318.09	5000	663.93	384	$1 \times 1 \times 1$	0.00	0.00	0.00	Solid
659.04	9000	721.66	192	$1 \times 1 \times 1$	0.00	0.00	0.12	SI
659.04	9000	719.64	192	$2 \times 2 \times 2$	0.00	0.00	0.10	SI
1318.09	9000	721.45	384	$1 \times 1 \times 1$	0.00	0.00	0.08	SI
659.04	9000	820.87	192	$1 \times 1 \times 1$	1.08	1.48	3.37	Fluid
659.04	12000	822.06	192	$2 \times 2 \times 2$	1.07	1.51	4.02	Fluid
1318.09	12000	820.75	384	$1 \times 1 \times 1$	1.06	1.40	3.98	Fluid

Table III. Structural information for HeSiO₂ compounds at 1000 GPa.

Space group	Lattice parameters (Å)	Atomic coordinates (fractional)			
		Si	O	O	He
<i>Pnma</i> -I	a = 7.3530	0.6125	0.7500	0.2546	
	b = 2.2022	0.9667	0.2500	0.2499	
	c = 2.8982	0.2153	0.2500	0.0276	
<i>Pnma</i> -II	He	0.3531	0.7500	0.0657	
	a = 7.4130	0.8618	0.7500	0.5341	
	b = 2.1865	0.0339	0.7500	0.2387	
	c = 2.8849	0.2835	0.7500	0.4700	
<i>Pnma</i> -III	He	0.1023	0.7500	0.7868	
	a = 3.9635	0.8139	0.7500	0.1580	
	b = 2.1028	0.6865	0.7500	0.4528	
	c = 5.5564	0.9774	0.7500	0.7208	
<i>Pnma</i> -IV	He	0.4570	0.2500	0.9751	
	a = 5.4683	0.3271	0.2500	0.1362	
	b = 2.1832	0.3930	0.7500	0.8795	
	c = 3.9197	0.6083	0.2500	0.6175	
<i>Pmn2</i> ₁	He	0.6496	0.7500	0.3703	
	a = 2.2011	0.0000	0.7249	0.7610	
	b = 3.6785	0.0000	0.0697	0.4804	
	c = 2.8973	0.5000	0.4334	0.7582	
<i>P2</i> ₁ /m	He	0.5000	0.7940	0.4424	
	a = 2.8890	0.3805	0.7500	0.7236	
	b = 2.1869	0.2124	0.2500	0.4332	
	c = 4.0453	0.7235	0.2500	0.9325	
	He	0.0984	0.2500	0.7953	

Table V. Structural information for HeSiO₄ compounds at 1000 GPa.

Space group	Lattice parameters (Å)	Atomic coordinates (fractional)		
		Si	O	He
<i>I422</i>	a = 3.2944	0.5000	0.5000	0.5000
		0.6312	0.8688	0.7500
		0.5000	0.5000	0.0000

Reference

1. Grimme, S., Ehrlich, S. & Goerigk, L. Effect of the damping function in dispersion corrected density functional theory. *J. Comput. Chem.* **32**, 1456–1465 (2011).
2. Blaha, P., Schwarz, K., Kvasnicka, D. & Luitz, J. WIEN2k: An Augmented Plane Wave Plus Local Orbitals Program for Calculating Crystal Properties (Karlheinz Schwarz, Technische Universitaet Wien, Vienna, 2001).
3. Klimeš, J., Bowler, D. R. & Michaelides, A. Chemical accuracy for the van der Waals density functional. *J. Phys. Condens. Matter* **22**, 022201 (2009).
4. Hamada, I. van der Waals density functional made accurate. *Phys. Rev. B* **89**, 121103 (2014).
5. Clark, S. J. *et al.* First principles methods using CASTEP. *Z. Für Krist. - Cryst. Mater.* **220**, 567–570 (2009).