

Observation of Tunneling Gap in Epitaxial Ultrathin Films of Pyrite-Type Copper Disulfide *

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We report scanning tunneling microscopy investigation on epitaxial ultrathin films of pyrite-type copper disulfide. Layer-by-layer growth of CuS_2 films with a preferential orientation of (111) on $\text{SrTiO}_3(001)$ and $\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8+\delta}$ substrates is achieved by molecular beam epitaxy growth. For ultrathin films on both kinds of substrates, we observe symmetric tunneling gap around the Fermi level that persists up to $\sim 15\text{K}$. The tunneling gap degrades with either increasing temperature or increasing thickness, suggesting new matter states at the extreme two-dimensional limit.

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Engineering of heterostructures and ultrathin films at the extreme of two-dimensional (2D) limit is a vibrant frontier in realizing novel electronic states of matter. The most notable recent breakthrough includes the single-layer variants of various transition-metal dichalcogenides^[1,2] and the monolayer FeSe films grown on several $\text{TiO}_{2-\delta}$ terminated substrates,^[3–5] which exhibit exotic properties distinct from their bulk counterparts, therefore have ignited explosive investigations. Aiming at search for new interfacial systems or ultrathin films that exhibit diverse and novel properties, we choose CuS_2 , a pyrite-type transition metal dichalcogenide, as a new specimen.

The pyrite-type transition metal dichalcogenides MX_2 , with $M = \text{Fe}, \text{Co}, \text{Ni}, \text{Cu}, \text{Zn}$ and $X = \text{S}, \text{Se}, \text{Te}$ (no ZnTe_2), show a large variety of electronic, magnetic and optical properties. Among the pyrites known so far, copper pyrites, CuS_2 , CuSe_2 and CuTe_2 are the only superconducting compounds.^[6] The pyrite-type CuS_2 , which can be obtained by synthesis under high pressures, shows metallic behavior associated with a second-order phase transition at 150 K, weak ferromagnetism with a Curie temperature of 23 K,^[7,8] and superconductivity at temperatures below $T_c = 1.5\text{K}$.^[9,10] Thus, there is possible coexistence of the superconductivity and the weak ferromagnetism below T_c . On the other hand, the previous transport, magnetization and specific heat measurements showed that there is no clear indication of strong electron correlations in the electronic properties of Cu pyrites.^[11] It is chalcogen p character rather than d characteristic of Cu that dominates at the Fermi level,^[11,12] which is in remarkable contrast to other $3d$ transition metal pyrites.

To explore the diverse and novel properties in ultrathin films of CuS_2 , in this work, we perform

molecular beam epitaxy (MBE) growth and scanning tunneling microscopy/spectroscopy (STM/STS) study of ultrathin films of CuS_2 on $\text{SrTiO}_3(001)$ and $\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_{8+\delta}$ (Bi-2212) substrates. Under a moderate S-rich condition, the single-phase pyrite CuS_2 grows in a layer-by-layer mode with a preferential orientation of (111). At higher S flux, CuS_2 islands with various heights instead of uniform films form, which conceives that S flux is a key factor for the growth of pyrite CuS_2 . Another interesting finding is symmetric tunneling gap with weak coherence peaks around the Fermi level, which occurs in ultrathin films and degrades with either increasing temperature or increasing thickness. The observation of such a tunneling gap suggests new matter states in ultrathin films of CuS_2 , which deserves further study.

Our experiments were conducted in a Createc ultrahigh vacuum STM system equipped with an MBE chamber for *in situ* sample preparation. The base pressure of the system is better than 1.0×10^{-10} Torr. The samples were immediately transferred into the STM head for data collection at 4.6 K after MBE growth. A polycrystalline PtIr tip was used throughout the experiments. STM topographic images were acquired in a constant current mode, with the bias voltage V_s applied to the sample. Tunneling spectra were measured by disabling the feedback circuit, sweeping the sample voltage V_s , and then extracting the differential conductance dI/dV using a standard lock-in technique with a small bias modulation of 673.2 Hz.

We chose TiO_2 -terminated Nb-doped (0.05 wt%) $\text{SrTiO}_3(001)$ obtained after heating to 1100°C and new cleaved optimal-doped Bi-2212 as substrates. Because of the extremely high vapor pressure of the S beam, the re-evaporation of S is a major concern especially in the precisely controlled growth of ultrathin

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CuS₂ films at the extreme 2D limit. In order to improve the sticking efficiency of S and the controllability of the chemical composition, we used the CuS compound, which decomposes to solid Cu₇S₄ and S₂ vapor when heated,^[13] as a S source and set a comparatively low substrate temperature of 80°C. At such low substrate temperature, the surface migration of the arrived atoms would become less effective. In order to compensate this problem, we lowered the growth rate to ~0.05 monolayer (ML)/min, accomplished by co-evaporating high-purity Cu (99.9999%) and CuS (99.98%) by Knudsen diffusion cells at cell temperatures of 850°C and 170°C, respectively. Here 1 ML refers to a nominal coverage which completely covers the SrTiO₃ or Bi-2212 surface. Under these conditions, the CuS₂ films grew in the layer-by-layer mode and showed good crystal quality with a preferential surface orientation of (111), as will be shown later on.

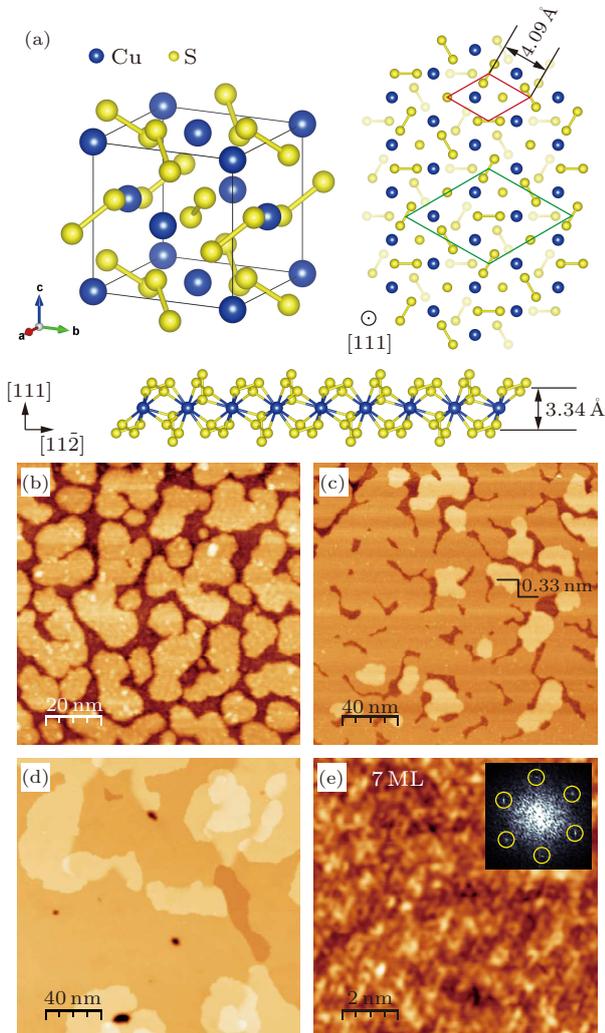


Fig. 1. (a) Crystal structure of pyrite CuS₂. (b)–(e) STM topographic images of CuS₂ on SrTiO₃(001) at various nominal coverages: (b) 0.7 ML ($V_s = 5$ V, $I = 50$ pA), (c) 1 ML ($V_s = 4.5$ V, $I = 40$ pA), and (d)–(e) 7 ML ((d) $V_s = 6$ V, $I = 50$ pA and (e) $V_s = 0.4$ V, $I = 100$ pA). The inset in (e) is the corresponding FFT images with the yellow circles marking the 1×1 Bragg points.

taining interpenetrating face-centered-cubic arrays of Cu²⁺ cations and S₂²⁻ anion dimers, with the lattice constant of 5.7898 Å.^[12] Within a unit cell, the Cu²⁺ cations occupy the face centers of the cubic cell, while the S₂²⁻ dimers are centered near the anion positions, as shown in upper left panel of Fig. 1(a). For the closest-packed (111) plane, as shown in right panel of Fig. 1(a), the surface exhibits a hexagonal structure, with an in-plane lattice constant of 8.19 Å, and each primitive unit cell (shown as a green rhombus) consists of four Cu²⁺ cations and four S₂²⁻ dimers. As schematically illustrated in bottom panel of Fig. 1(a), along the (111) direction, three of the four S₂²⁻ dimers are tilted with respect to the (111) orientation, while the remaining S₂²⁻ dimer is along the (111) direction. Therefore, the eight S atoms, separated into four different planes, are sandwiched between the two adjacent Cu layers with a distance of 3.34 Å. That is, the pyrite CuS₂ forms a quasi-layered structure along the (111) direction. For convenience, we hereafter introduce a reduced unit cell (including one Cu atom) marked by the red rhombus in upper right panel of Fig. 1(a). In the case of substrates, the TiO₂ termination layer for the SrTiO₃(001) and BiO termination layer for Bi-2212 have tetragonal structures with lattice constant of 3.9 Å and 3.8 Å, respectively.

Despite the structure mismatch between hexagonal in-plane structure of CuS₂ and tetragonal in-plane structures in either SrTiO₃(001) or Bi-2212 substrates, crystalline CuS₂ films with atomically flat (111) surface form on such two substrates. Figures 1(b)–1(e) depict a series of typical STM topographic images of CuS₂ on SrTiO₃(001) at various coverages. Initially, at 0.7 ML, some irregular flat CuS₂ patches with uniform height distribute on the surface, as shown in Fig. 1(b). The apparent thickness of the CuS₂ patches is bias-dependent, increasing from 5.4 Å to 7.0 Å with sample bias decreasing from 4.5 V to 3 V, due to contrast between the density of states of CuS₂ and SrTiO₃ substrates. With coverage increased to 1 ML (Fig. 1(c)), the CuS₂ patches coalesce and form continuous films of 1 ML in thickness. Some second layer patches occur occasionally on the first layer CuS₂ films with constant thickness of 3.3 Å, exactly the same as the layer distance along the (111) direction shown in Fig. 1(a), indicating a nearly layer-by-layer growth. This layer-by-layer growth continues as the coverage increases. For example, at the nominal coverage of 7 ML, uniform and continuous film with atomically flat surface forms, except for small amount of redundant patches above that (Fig. 1(d)). Shown in Fig. 1(e) is a zoom-in image taken on the flat terrace of 7 ML films. Although long-range ordered structure is hardly identified directly from topographic image, the corresponding fast Fourier transformation (FFT) image (inset of Fig. 1(e)) indicates hexagonal structure with lattice period of 4.0 Å, which is half the in-plane lattice constant of the CuS₂(111) plane and agrees excellently with the nearest Cu–Cu or S₂–S₂ distance

Pyrite CuS₂ has a rock salt type structure, con-

depicted in upper right panel of Fig. 1(a). Considering that chalcogen p character rather than d characteristic of Cu dominantly contributes the density of states around the Fermi level,^[11] we attribute the atomic periodic structure observed here to S_2^{2-} dimers. Combining the layer thicknesses and in-plane structure, we conclude that single-phase (111)-oriented pyrite CuS_2 thin films with S_2 termination form on the $SrTiO_3(001)$ substrate.

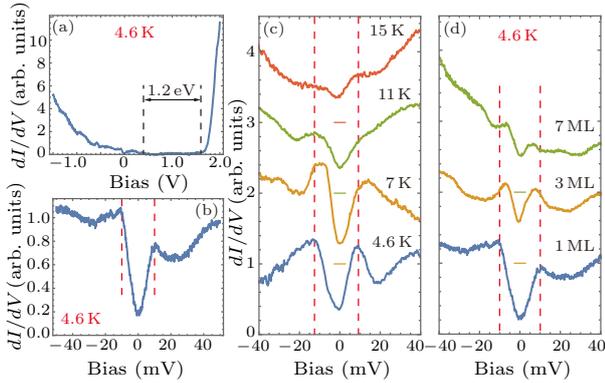


Fig. 2. (a) and (b) dI/dV tunneling spectra taken on 1 ML CuS_2 films on $SrTiO_3(001)$ ((a) $V_s = 2$ V, $I = 100$ pA; (b) $V_s = -50$ mV, $I = 100$ pA). (c) Tunneling spectra taken on 1 ML films at various temperatures ($V_s = 60$ mV, $I = 100$ pA). (d) Tunneling spectra taken on films with various thickness: 1 ML ($V_s = -50$ mV, $I = 100$ pA), 3 ML ($V_s = -60$ mV, $I = 200$ pA) and 7 ML ($V_s = 60$ mV, $I = 100$ pA). The spectra in (c) and (d) are shifted along the vertical axis. The horizontal bars indicate the zero-conductance position of each curve. The black dashed lines in (a) show the band gap. The red dashed lines in (b)–(d) are guide for eyes, showing the change of tunneling gap.

The most interesting finding in this work is the observation of symmetric tunneling gap around the Fermi level. Figure 2 summarizes the differential tunneling spectra (dI/dV) taken on CuS_2 films of various thickness and at various temperatures. Displayed in Fig. 2(a) and 2(b) are dI/dV spectra of 1 ML CuS_2 taken at 4.6 K in a large bias range (from -1.5 V to 2.0 V) and small bias range (from -50 mV to 50 mV), respectively. We observed a band gap of 1.2 eV with the valence band top located at 0.4 eV above the Fermi level (Fig. 2(a)). The gap value is consistent with band structure calculation result on bulk CuS_2 , however the gap position moves ~ 0.4 eV closer to the Fermi level.^[12] This shift suggests that the 1 ML CuS_2 films here are electron-doped compared with their bulk counterparts. This is most likely due to the charge transfer from the underlying $SrTiO_3$ substrates which act as charge reservoir, as previously revealed in $FeSe/SrTiO_3$ ^[3–5] and $LaAlO_3/SrTiO_3$ ^[14] hetero-structures. More striking is that a symmetric gap opens around the Fermi level, characterized by weak coherence peaks at ± 10 meV and nonzero zero-bias-conductance (ZBC) (Fig. 2(b)). With increasing temperature up to 15 K, the gap degrades gradually in magnitude and ZBC increases simultaneously (Fig. 2(c)). This kind of tunneling gaps keep

up for films at least thinner than 7 ML and exhibit thickness-dependent behaviors. For comparison, we summarized the typical dI/dV spectra of 1 ML, 3 ML and 7 ML CuS_2 films in Fig. 2(d). Clearly, the magnitude of gap decreases and the ZBC increases with increasing thickness, suggesting a low dimensional or interfacial origin of such tunneling gaps.

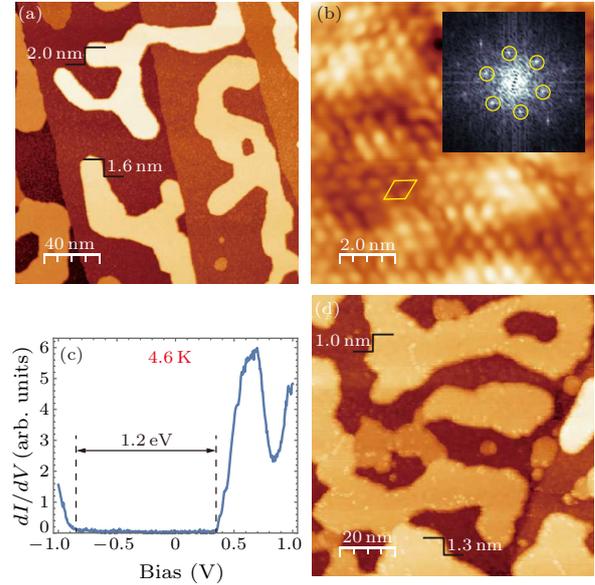


Fig. 3. (a)–(b) Topographic image of 1 ML CuS_2 after being annealed at 200°C ((a) $V_s = 2.5$ V, $I = 50$ pA; (b) $V_s = 1.3$ V, $I = 200$ pA). In (b), the yellow rhombus labels the unit cell. The inset is the corresponding FFT image showing reconstruction, as labelled by the yellow circles. (c) dI/dV tunneling spectra taken on CuS_2 islands on $SrTiO_3(001)$ ($V_s = 1$ V, $I = 100$ pA). The black dashed lines show the band gap. (d) Topographic image of 1 ML CuS_2 after being annealed under S flux at 60°C ($V_s = 3$ V, $I = 50$ pA).

To modulate and further explore the properties of the CuS_2 films at the 2D limit, we carried out the *in situ* post-annealing of as-grown 1 ML CuS_2 films. As shown in Fig. 3(a), with annealing at 200°C, the films evolve to striped-islands with heights increased to 1.6–2.0 nm, corresponding to 4–5 ML. Some of the striped islands crossed $SrTiO_3$ steps, indicating the long-range diffusion of CuS_2 . Such morphology change is accompanied by remarkable surface reconstruction and electronic properties change. Instead of 1×1 lattice, the surface exhibits a $\sqrt{3} \times \sqrt{3}$ -R30° reconstruction, as identified directly from the periodicity on the real space topographic image and the corresponding FFT image in Fig. 3(b). Figure 3(c) typifies the differential conductance dI/dV spectra of $\sqrt{3} \times \sqrt{3}$ -R30° surface. Compared with the 1×1 lattice, the semiconducting gap of 1.2 eV remains but shifts downwards and crosses the Fermi level. It seems that the CuS_2 films evolve to semiconducting states with the formation of $\sqrt{3} \times \sqrt{3}$ -R30° reconstruction from metallic states with 1×1 lattice. Our previous work on pyrite $CuSe_2$ showed that the chalcogenide concentration might lead to surface relaxation and thus various surface reconstructions,^[15] hence we speculate that

the above results observed on CuS_2 islands could be due to the S desorption with post-annealing. The desorption of S will lead to electron doping, which agrees with the upward Fermi level shift shown in Fig. 3(c) compared with Fig. 2(a). It is worth noting that the spectra taken on the isolated CuS_2 islands may be affected by the possible non-ohmic contact between CuS_2 and SrTiO_3 substrates that the tunneling current passes through. To quantitatively characterize the difference between $\sqrt{3}\times\sqrt{3}$ and 1×1 phases, comparing the spectra taken on the individual phases but in the same form of continuous films is required.

We also conducted annealing under S flux, and only found the films evolved to higher islands even under very low annealing temperature of 60°C (Fig. 3(d)). This remarkable morphology variation, together with the Volmer–Weber growth at higher S flux (not shown), suggests that S flux plays an important role in the growth mode of CuS_2 films, which is in sharp contrast with the growth of other chalcogenides such as selenides.^[3–5,15] CuS_2 islands prepared under higher S flux show similar spectral features as those formed after annealing under S flux, which again indicates the susceptibility of electronic properties to S. This extra susceptibility naturally results in spatially inhomogeneous properties, as we observed spatially inhomogeneous spectra at each states. Here, we mainly show the most characteristic and reproducible ones.

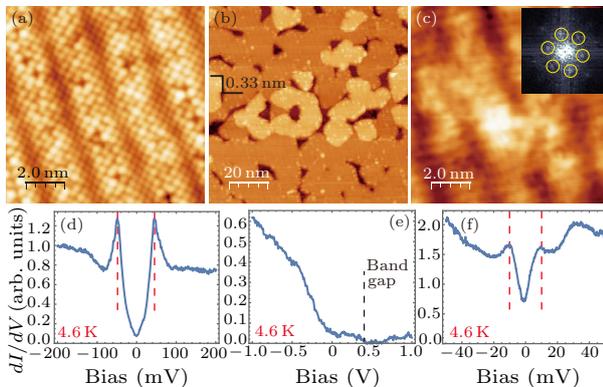


Fig. 4. (a) Topographic image of as-cleaved Bi-2212 ($V_s = 0.6\text{ V}$, $I = 100\text{ pA}$). (b) and (c) Topographic images of CuS_2 grown on Bi-2212 with nominal coverage of 1 ML ((b) $V_s = 3\text{ V}$, $I = 50\text{ pA}$; (c) $V_s = 0.5\text{ V}$, $I = 100\text{ pA}$). The inset of (c) is the FFT image showing reconstruction as labeled by the yellow circles. (d) Tunneling spectrum taken on exposed Bi-2212 after the growth of CuS_2 ($V_s = -0.2\text{ V}$, $I = 100\text{ pA}$). (e) and (f) Tunneling spectra taken on 1 ML CuS_2 on Bi-2212 ((e) $V_s = -1\text{ V}$, $I = 50\text{ pA}$; (f) $V_s = -50\text{ mV}$, $I = 100\text{ pA}$). The red dashed lines in (d) and (f) show coherence peaks of the tunneling gaps.

To identify the tunneling gap shown in Fig. 2 is mainly due to intrinsic low dimensional or extrinsic interfacial effect, we grew CuS_2 films on another substrate—the BiO surface of the new cleaved Bi-2212 crystals. As displayed in Fig. 4(a), the BiO surface shows the characteristic b -axis supermodulation with a period of 26 \AA and square lattice with a lattice con-

stant of 3.8 \AA . On such BiO surface, the CuS_2 grows via the same layer-by-layer mode and the step height is 3.3 \AA (Fig. 4(b)). On the surface of 1 ML CuS_2 , as shown in Fig. 4(c), the b -axis supermodulation of BiO surface still can be seen clearly. Despite that, the FFT image shows hexagonal lattice with periodic constants of 6.9 \AA , which agrees well with that of $\sqrt{3}\times\sqrt{3}\text{-R}30^\circ$ reconstruction. Similar to CuS_2 films on $\text{SrTiO}_3(001)$ substrates, the Fermi level lies at $\sim 0.4\text{ eV}$ below the valence band top (Fig. 4(e)) and a tunneling gap with weak coherence peaks at $\pm 10\text{ meV}$ occurs around the Fermi level (Fig. 4(f)). On the other hand, in the exposed BiO regions, the surface structure looks the same as the new-cleaved BiO surface and the differential conductance dI/dV spectra also show the characterized pseudo-gap of 45 meV (Fig. 4(d)), which hints sharp interface between CuS_2 films and BiO layer.

We compare the morphologies and spectra of CuS_2 films on $\text{SrTiO}_3(001)$ and Bi-2212 substrates and summarize our results as below. (1) The band gap position, as identified from the spectra shown in Figs. 2(a) and 4(e), moves $\sim 0.4\text{ eV}$ closer to Fermi level compared with the calculation results for bulk CuS_2 .^[12] This suggests that both SrTiO_3 and Bi-2212 substrates could introduce electron doping to CuS_2 films, while low-dimensional effect could also contribute to LDOS change. (2) Despite various interfaces, symmetric tunneling gap with magnitude of $\sim 10\text{ meV}$ can be consistently observed on both $\text{SrTiO}_3(001)$ and Bi-2212 substrates, which suggests intrinsic novel properties in the ultrathin films of CuS_2 . (3) The tunneling gap degrades with either increasing thickness or increasing temperature, suggesting new matter states in ultrathin CuS_2 films that become strongest at the extreme 2D limit. We did not see signature of charge density waves. In contrast to Coulomb gap previously observed on semiconducting films,^[16] here the relatively large tunneling gap is observed on metallic state. Although no noticeable strong electron correlation has been shown in its bulk counterparts,^[11] the ultrathin films of CuS_2 with interfacial doping could probably exhibit exotic properties. The tunneling gap observed on CuS_2 films exhibit similar features to the pseudo-gap in cuprate, as clearly seen from the comparison between the spectra shown in Figs. 2(b)–(d), 4(f) and the spectrum taken on BiO surface shown in Fig. 4(d). The origin of the tunneling gap deserves further study. (4) The contrasting morphologies and surface reconstructions, i.e., films vs islands and metallic 1×1 vs semiconducting $\sqrt{3}\times\sqrt{3}$, can be both tuned by S flux. Similarly, the electronic properties are susceptible to S flux, which is a significant feature of sulfide compound. This requires fine controllability on the sample stoichiometry in study on such sulfide compounds.

In summary, we successfully prepared single crystalline films of CuS_2 by using MBE and observed tunneling gap with weak coherence peaks in ultrathin films at the extreme 2D limit. The method of MBE preparation of ultrathin CuS_2 films could be used for

other transition metal disulfides, such as MoS₂ and TaS₂, which exhibit diverse interesting features of optical and electrical properties and are being hotly studied. Compared with exfoliation and chemical vapor deposition methods that previously used, the MBE growth guarantees single phase crystalline films with fine stoichiometry controllability. The symmetric tunneling gap with weak coherence peaks around Fermi level suggests new matter states in ultrathin films of CuS₂, which deserves further study.

References

- [1] Xu X, Yao W, Xiao D and Heinz T F 2014 *Nat. Phys.* **10** 343
- [2] Navarro-Moratalla E, Island J O, Mañas Valero S, Pinilla-Cienfuegos E, Castellanos-Gomez A, Quereda J, Rubio-Bollinger G, Chirolli L, Silva-Guillen J A, Agrait N, Steele G A, Guinea F, van der Zant H S and Coronado E 2016 *Nat. Commun.* **7** 11043
- [3] Wang Q Y, Li Z, Zhang W H, Zhang Z C, Zhang J S, Li W, Ding H, Ou Y B, Deng P, Chang K, Wen J, Song C L, He K, Jia J F, Ji S H, Wang Y Y, Wang L, Chen X, Ma X C and Xue Q K 2012 *Chin. Phys. Lett.* **29** 037402
- [4] Wang L, Ma X C and Xue Q K 2016 *Supercond. Sci. Technol.* **29** 123001
- [5] Wang L and Xue Q K 2017 *AAPPS Bull.* **27** 3
- [6] Bither T A, Bouchard R J, Cloud W H, Dokohue P C and Siemoks W J 1968 *Inorg. Chem.* **7** 2208
- [7] Krill G, Panissod P, Lapierre M F, Gautier F, Robert C and Eddinetg M N 1976 *J. Phys. C* **9** 1521
- [8] Kontani M, Tutui T, Moriwaka T and Mizukoshi T 2000 *Physica B* **284-288** 675
- [9] Bither T A, Prewitt C T, Gillson J L, Bierstedt P E, Flippen R B and Young H S 1966 *Solid State Commun.* **4** 533
- [10] Munson R A, DeSorbo W and Kouvel J S 1967 *J. Chem. Phys.* **47** 1769
- [11] Ueda H, Nohara M, Kitazawa K, Takagi H, Fujimori A, Mizokawa T and Yagi T 2002 *Phys. Rev. B* **65** 155104
- [12] Hou Z, Li A, Zhu Z and Huang M 2004 *J. Mater. Sci. Technol.* **20** 429
- [13] Brunetti B, Piacente V and Scardala P 1994 *J. Alloy. Compd.* **206** 113
- [14] Ohtomo A and Hwang H 2004 *Nature* **427** 423
- [15] Peng J P, Zhang H M, Song C L, Jiang Y P, Wang L, He K, Xue Q K and Ma X C 2015 *Chin. Phys. Lett.* **32** 068104
- [16] Butko V Y, DiTusa J F and Adams P W 2000 *Phys. Rev. Lett.* **84** 1543